Al - (Al9Co2, Al13Co4) Powder Metallurgy Processed Composite Materials: Analysis of Microstructure, Sliding Wear and Aqueous Corrosion

G. Bakoulis, Assist.Prof AE Karantzalis Prof. AG Lekatou, Phd, A.Poulia, Phd AK Sfikas, Phd K Lentzaris

Laboratory of Applied Metallurgy Dept of Material Science and Engineering, Uoi

1.ABSTRACT

The present effort was undertaken in order to investigate the manufacturing of aluminum based composites, reinforced by a dual intermetallic phase, Al₉Co₂ -Al₁₃Co₄, using a powder metallurgy route. The metal matrix was reinforced with compositions 2, 5 and 10% vol. The manufacturing of the dual Al₉Co₂ - Al₁₃Co₄ powders was carried out by vacuum arc melting followed by milling to produce finesized powders. The composite materials were produced by a common powder metallurgy route, including mixing in a mortar, consolidation by cold pressing, and sintering under vacuum at 600°C. The wear properties of the composite were examined by dry sliding wear against an alumina counter face at 10cm/s, a load at 1N and a distance at 1000m. Macro hardness was also assessed. Aqueous corrosion experiments in 3.5% NaCl at 25°C were carried out in all three different compositions. Optical microscopy and SEM-EDX analysis were used in order to ascertain the microstructure, the sliding wear tacks and debris and the corroded surfaces. The sliding wear phenomena were related to the microstructural features and approached based on classic theories of Al composite wear behaviour and so was the corrosion response of the produced materials.

2. INTRODUCTION

Aluminium Matrix Composites (AMCs) have gained intensive research interest during the last four decades, due to their improved mechanical and physical properties and have been considered as potential candidate materials for various, aerospace, automotive, structural and other civil and military applications [1-5]. Especially, AMCs have found application in the manufacture of various automotive and aerospace, engine or structural components such as cylinder blocks, pistons, piston insert rings, aircraft empennage, fuselage in fighter aircraft, rotary aircraft were dry sliding wear and corrosion resistance are predominant properties [6,7].

Various production methods for AMCs have been developed within these years, with the casting based processes being at the forefront due to the ease-to-handle operation scheme and, most importantly, their relatively low cost. Powder metallurgy, nevertheless, despite their relatively high cost, has attracted great attention in the manufacture of AMCs due to the unique properties these production routes may provide [1-5,8-12].

3. EXPERIMENTAL PROCEDURE

Appropriate amounts, determined by the Al-Co phase diagram [13], of Al (<325 mesh) and Co powders (<40 μ m) both by Sigma-Aldrich were weighted and mixed targeting the composition corresponding to the Al₉Co₂ intermetallic phase. Small loads (5g) of the mixture were mechanically pressed under 300 bar to form coupons that there were arc melted in an arc melting furnace (VAM) using W electrode at 120A current using Ar as protecting atmosphere. The resulting droplets were crushed and subjected to planetary ball milling for 10 minutes with the presence of alcohol. The resulting powder was dried and then mixed with appropriate amount of Al powder in order to produce 2, 5 and 10 vol.% in Al₉Co₂ composite materials respectively. Coupons of approximately 5g were formed by mechanical pressing under 250 bar and subjected to free sintering for 9h at 600°C under Ar atmosphere.

Specimens were metallographically prepared and examined under optical microscope (Leica 4000DM) and SEM-EDX (Jeol 6510 LV, x-Act Oxford Instruments). Sliding wear experiments were conducted using a CSM ball on disk tribometer, with 6mm Al₂O₃ ball as the counter-body, at 10 cm/s linear speed, 5mm rotation diameter, 20Hz acquisition time, 1N externally applied load and 1 km as overall sliding distance. The tests were interrupted every 200m for weighting the samples and record the resulting mass loss. Wear tracks and debris were also examined with SEM-EDX. Hardness measurements were conducted using a Universal Hardness Tester 700M by Innovatest.

Aqueous corrosion experiments were conducted in properly prepared samples using a Gill AC galvanostat by ACM Instruments, within a 3.5% NaCl solution, using an Ag/AgCl/ 3.5 KCl reference electrode. The experiments were conducted keeping the solution pH at 7, at scanning rate of 10mV/min within the range of 1000 to 1500 mV after having ascertained the rest potential for 2h.

4. RESULTS AND DISCUSSION

4.1 MICROSTRUCTURAL OBSERVATIONS

Figure 1 shows a panoramic view of the precursor Al-Co powder after vacuum arc melting and milling. It can be observed that the powder size lies within the range of approximately 1-50 µm. In the same Figure 1, the XRD analysis of the Al-Co intermetallic powder shows the existence of the Al₉Co₂ phase - as expected form the initial targeted composition – along with the presence of Al and Al₁₃Co₄. A possible solidification sequence for the precursor Al-Co powder may be as follows: (a) The high temperature achieved during arc melting ensures entire melting of the raw materials, (b) According to the Al-Co phase diagram, at ~1100°C, Al₃Co solidification starts. At 1093°C, peritectic reaction of the Al-rich liquid with Al₃Co occurs, leading to the formation of Al₁₃Co₄, (c) At 970°C, Al₁₃Co₄ peritectically reacts with molten Al to form Al₉Co₂ and (d) However, according to the Al-Co equilibrium phase diagram [13], the Al₁₃Co₄ phase should not exist at ambient temperatures. All the same, the high cooling rates during solidification following arc melting, do not allow for completion of the diffusion processes, and, consequently, attainment of an equilibrium state. Hence, the Al (l) + Al₁₃Co₄ = Al₉Co₂ peritectic reaction cannot run into completion, allowing the Al₁₃Co₄ phase to persist. Actually, its extensive presence at room temperature manifests the high cooling rates involved in the arc melting process adopted. To sum up, the formation of Al₁₃Co₄ is most likely attributed to the relatively high cooling rates during solidification after vacuum arc melting that most likely does not permit the $Al_{13}Co_4 \rightarrow Al_9Co_2$ transformation completion.



Figure 1: SEM micrograph showing the shape and size of the precursor Al-Co powder and XRD analysis showing the coexistence of both Al₉Co₂ and Al₁₃Co₄ phases

Figure 2 presents optical micrographs of the different produced composites. It can be observed that in all cases the microstructure is characterized by the uniform particle distribution either in the form of isolated particles or particle clusters. The existence of residual porosity is also evident. The microstructure of the produced materials is presented more clearly under SEM examination, as shown in Figure 3. The cross sections of the produced composites (Figure 3) reveal the uniform particle distribution. It is also evident the presence of particle clusters and residual porosity that both increase with increasing the reinforcement content. The clustering increase is most likely associated with primary particle clusters that did not disintegrate during the milling process.



Figure 2: Optical micrographs showing the microstructure of the final composite materials



Figure 3: SEM images showing the cross sections of the produced materials, enlightening further their morphological features

The increased porosity, on the other hand, maybe the result of either the observed particle clustering (particle clusters contain voids and gaps that can evolve into porosity upon processing) or insufficient sintering (the higher the reinforcing particles the more obstacles for their surrounding Al grains to get into intimate contact in order to promote and sustain effective sintering. An interesting observation concerning the phase composition of the final composites arises from the elemental analysis on various reinforcing particles for all the different compositions, as presented in Figure 4. It can be seen that the Al-Co ratio in most of the cases, remains constant and practically corresponds to the Al₉Co₂ phase stoichiometry. Such observation, most likely suggests that upon the prolonged sintering conditions (600°C for 9h) allows the off equilibrium Al₁₃Co₄ phase to be fully transformed into Al₉Co₂ intermetallic compound.

4.2 SLIDING WEAR BEHAVIOUR

Figure 5 shows the volume loss as a function of the sliding distance for the different produced composites and their corresponding wear rates along with their hardness values. It can be observed that the wear rate is decreased with increasing the reinforcing particle content. It is also evident that the higher the particle content the higher the hardness value. These observations clearly verify the beneficial action of the reinforcing phase on the sliding wear resistance. Similar conclusions have been drawn by other researchers [17].



Figure 4: Elemental analysis on various different reinforcing particles, reveling that the Al₉Co₂ phase prevails after the completion of the sintering process.

The wear resistance improvement is most likely associated with: a) the fact that reinforcing particles support the externally applied load, restricting in such way the intimate contact between the soft matrix and the counterbody which is responsible for the extreme plastic deformation and the resulting degradation phenomena, b) the provision of thermal, to the matrix stability, that postpones potential softening phenomena that promote severe plastic deformation and c) strain hardening effect due to particle dispersion strengthening effect that improves the matrix resistance to plastic deformation. All these potential improvement mechanisms are also well documented in other experimental works [14-17]. It has also to be mentioned that the wear response of the produced materials are in line with their hardness and the classic theory of Archard [21]: the higher the hardness the higher the wear resistance.



Figure 5: Volume loss vs sliding distance and corresponding wear rates of the produced materials

A panoramic view of the wear track morphology of the produced composites is presented in Figure 6. It can be observed the characteristic "valley – hill" morphology - a result of the strain hardening effect of the soft matrix – which Sharkar [22] has explained in details the formation of and other research efforts also verified [17]. SEM-EDX point analysis reveals the presence of both oxide phases and cracks on the wear track surfaces. Both these features are responsible for the initiation and the promotion of the degradation phenomena which practically involves the development of cracks due to plastic deformation and or thermal fatigue on the oxide layer that eventually are promoted, expanded, bridged and lead to material loss [14-17]. The generated upon the sliding action debris is shown in Figure 7. This debris is also responsible for the wear phenomena especially for the slight abrasive action. It can be observed that the debris become finer and more equiaxed with the increase of sliding distance and the particle volume fraction. The sliding distance leads to prolonged surface oxide phases formation, which are brittle and are continuously entrapped at the counterfaces' gap and gradually fragmented into fine particles. The high particle content on the other hand increases the brittleness of the material, reducing its plasticity and leads to the formation of finer and more equiaxed debris [21].



Figure 6: Panoramic view of the wear track morphology along with EDX analysis showing the presence of oxide phases. Crack formation is also evident.



Figure 7: Debris morphology as a function of the distance and the reinforcing particle content

4.3 AQUEOUS CORROSION BEHAVIOUR

Figure 8 shows the potentiodynamic polarization curves of the different produced materials and Figure 9 the forward polarization curves of these systems. Based on the data of these figures it can be postulated that: a) all the produced composites show a better behaviour to localized corrosion compared to that of the monolithic Al alloy (breakdown potential (Eb) Al < Eb composite) with that of the 10 vol% system being the most optimum one as far the pitting corrosion resistance is concerned b) after the energetic corrosion of stage 1, the monolithic alloy in stage 2 shows lower current densities compared to that of the composite materials due to the fact that the composite systems exhibit galvanic corrosion between the Al matrix and the Al₉Co₂ reinforcing phase and c) the steeper slopes of the composite materials in stages 3 and 4 is most likely a result of the greater extent of the mixed surface oxide phases consisting of Al and Co.

More specifically, in the case of the composite materials, in stage 2 the energetic corrosion is retarded due to the formation of either Al₂O₃ surface films of hydrated semiconducting Al(OOH) and Al(OH)₃ compounds. In stage 3, between the Al matrix and the Al₉Co₂ reinforcement, pitting corrosion takes place and in stage 4 a pseudo-passivation phenomenon can be observed as a result of possibly both the deposition of oxides and hydroxides within the pit cavities and/or the formation of mixed oxide and hydroxide phases of Al and Co. All these speculations are in agreement with previous research efforts [19,20].

5. CONCLUSIONS

 $Al - Al_9Co_2$ based composites were successfully produced by a powder metallurgy based route. The particle distribution was uniform with some degree of clustering and some porosity being observed, especially for the higher reinforcing particle concentrations.

Al₉Co₂ was identified as the only reinforcing phase due to the elimination of the primary Al₁₃Co₄ phase after the prolonged sintering stage.

Sliding wear resistance was increased with increasing the reinforcing particles volume fraction as did the hardness values, verifying the classic theories of aluminum metal matrix composites wear response. The familiar for these materials characteristic wear track landscape was also observed.

Aqueous corrosion experiments, showed that the composite systems have a better pitting corrosion resistance compared to that of the monolithic alloy with that of 10 vol.% reinforcing content, being the optimum one. A galvanic cell between the Al matrix and the nobler Al₉Co₂ is established and passivated at higher current densities.

The high sliding wear and localized corrosion resistance of the Al - 10 vol.% (Al₉Co₂, Al₁₃Co₄) constitute promising attribute for the future development of Al based composites with even higher reinforcement compositions, taking into account the desire for low weight, low cost of raw materials, high ductility and high fracture toughness.

6. ACKNOWLEDGEMENSTS

Authors would like to thank Prof. Sofoklis Makridis who applied us part of the arcmelting apparatus system under his patent https://www.researchgate.net/publication/200664162_Multiapparatus_Arc_Melting_f or_Rapid_Solidification_Processes?ev=prf_pub.

7. REFERENCES

- [1] LOYD DJ. "Particle reinforced aluminium and magnesium matrix composites." *International Materials Reviews*. 1994; 39(1):1-23. doi: http://dx.doi.org/10.1179/imr.1994.39.1.1.
- [2] ROHATGI PK, ASTHANA R and DAS S. "Solidification, structures, and properties of cast metal-ceramic particle composites." *Interna-tional Materials Review*. 1986; 31(1):115-139. doi: http://dx.doi. org/10.1179/imtr.1986.31.1.115.
- [3] MORTENSEN A and JIN I. "Solidification processing of metal matrix composites." *International Materials Review*. 1992; 37(1):101-128. doi: http://dx.doi.org/10.1179/imr.1992.37.1.101.
- [4] ROHATGI PK, RAY S, ASTHANA R andNARENDRANATH CS. "Interfaces in cast metal-matrix composites." *Materials Science and Engineering*: A. 1993; 162(1-2):163-174. doi: http://dx.doi.org/10.1016/0921-5093(90)90041-Z.
- [5] ASTHANA R and TEWARI SN. "Interfacial and capillary phenomena in solidification processing of metal-matrix composites." *Com-posites*

Manufacturing. 1993; 4(1):3-25. doi: http://dx.doi. org/10.1016/0956-7143(93)90012-W.

- [6] DEUIS RL, SUBRAMANIAN C and YELLUP JM. "Dry sliding wear of aluminium composites - A review". *Composites Science and Technology*. 1997; 57(4):415-435. doi: https://doi.org/10.1016/S0266-3538(96)00167-4.
- [7] ESAKKIRAJ ES, SURESH S, MOORTHI NSV, KUMAR MK and RANJITH SMJ. "Study of Mechanical Behaviour of Stir Cast Aluminium Based Composite Reinforced with Mechanically Ball Milled TiB₂ Nano Particles." Advanced Materials Research. 2014; 984-985:410-415. doi: https://doi.org/10.4028/www.scientific.net/AMR.984-985.410.
- [8] LIU YB, LIM SC, LU L and LAI MO. "Recent development in the fabrication of metal matrix-particulate composites using powder metal-lurgy techniques." *Journal of Materials Science*. 1994; 29(8):1999-2007. doi: http://dx.doi.org/10.1007/BF01154673.
- [9] LIU ZY, WANG QZ, XIAO BL, MA ZT and LIU Y. "Experimental and modeling investigation on SiC_p distribution in powder metallurgy processed SiC_p/2024 Al composites." *Materials Science and Engineering*: A. 2010; 527(21-22):5582-5591. doi: http://dx.doi.org/10.1016/j. msea.2010.05.006.
- [10] POURNADERI S, MAHDAVI S and AKHLAGHI F. "Fabrication of Al/ Al₂O₃ composites by in-situ powder metallurgy (IPM)." *Powder Technology*. 2012; 229:276-284. doi: http://dx.doi.org/10.1016/j. powtec.2012.06.056.
- [11] RAHIMIAN M, PARVIN N and EHSANI N. "The effect of production param-eters on microstructure and wear resistance of powder metallur-gy Al– Al₂O₃ composite." *Materials and Design*. 2011; 32(2):1031-1038. doi: http://dx.doi.org/10.1016/j.matdes.2010.07.016.
- [12] RAHIMIAN M, EHSANI N, PARVIN N and BAHARVANDI HR. "The effect of par-ticle size, sintering temperature and sintering time on the properties of Al–Al₂O₃ composites, made by powder metallurgy." *Journal of Materials Processing Technology*. 2009; 209(14):5387-5393. doi: http://dx.doi.org/10.1016/j.jmatprotec.2009.04.007.
- [13] CALLISTER AJ. "The Al-Co (Aluminum-Cobalt) system." Bulletin of Alloy Phase Diagrams. 1989; 10(6):646–650. doi: http://dx.doi. org/10.1007/BF02877635.
- [14] KUMAR N, GAUTAM G, GAUTAM RK, MOHAN A and MOHAN S. "Wear, friction and profilometer studies of in-situ AA5052/ZrB₂ composites."

Tribology International. 2016; 97:313–326. doi: http://dx.doi. org/10.1016/j.triboint.2016.01.036.

- [15] RENGASAMY NV, RAJKUMAR M and KUMARAN SS. "Mining environment applications on Al 4032-Zrb₂ and Tib₂ in-situ composites." *Journal of Alloys and Compounds.* 2016; 658:757-773. doi: http://dx.doi. org/10.1016/j.jallcom.2015.10.257.
- [16] GOUSIA V, TSIOUKIS A, LEKATOU A and KARANTZALIS AE. "Al-MoSi₂Composite Materials: Analysis of Microstructure, Sliding Wear, Solid Particle Erosion, and Aqueous Corrosion." *Journal of Materials Engineering and Performance*. 2016; 25(8):3107-3120. doi: http://dx.doi. org/10.1007/s11665-016-1947-1.
- [17] TROWSDALE AJ, NOBLE B, HARRIS SJ, GIBBINS ISR, THOMPSON GE and WOODS GC. "The influence of silicon carbide reinforcement on the pitting behavior of aluminum." *Corrosion Science*. 1996; 38(2):177–191. doi:http://dx.doi.org/10.1016/0010-938X(96)00098-4.
- [18] TAHAMTAN S, HALVAEE A, EMAMY M and ZABIHI MS. "Fabrication of Al/ A206–Al₂O₃ nano/micro composite by combining ball milling and stir casting technology." *Materials and Design.* 2013; 49:347-359. doi: http://dx.doi.org/10.1016/j.matdes.2013.01.032.
- [19] DE SALAZAR JMG, URENA A, MANZANEDO S and BARRENA MI. "Corrosion behaviour of AA6061 and AA7005 reinforced with Al₂O₃ par-ticles in aerated 3.5% chloride solutions: potentiodynamic mea-surements and microstructure evaluation." *Corrosion Science*. 1998; 41(3):529-545. doi: http://dx.doi.org/10.1016/S0010-938X(98)00135-8.
- [20] HOLLINGSWORTH EH and HUNSICKER HY. "Corrosion of aluminium and alu-minium alloys." In: Davis JR, editor. *ASM Handbook*, vol. 13: Corrosion, 9th ed. Materials Park, OH: ASM International; 1987. p.583–609.
- [21] ARCHARD JF. "Contact and Rubbing of Flat Surfaces." *Journal of Applied Physics*. 1953; 24(8):981-988. doi: http://dx.doi. org/10.1063/1.1721448.
- [22] SARKAR AD. "Friction and wear." Academic Press; 1980. p. 205-209

Παρασκευή σύνθετων υλικών μεταλλικής μήτρας αλουμινίου με ενίσχυση Al9Co2 με την χρήση κονεομεταλλουργικής οδού και μελέτη της συμπεριφοράς τους σε φθορά και διάβρωση.

Γ. Μπάκουλης, Αναπλ. Καθ. Α.Καράτζαλης Καθ. Α. Λεκάτου, Phd, Α.Πούλια, Phd Α. Σφήκας, Phd Κ Λέντζαρης

Εργαστήριο Εφαρμοσμένης Μεταλλουργίας Τμήμα μηχανικών επιστήμης και τεχνολογίας υλικών, Πανεπιστήμιο Ιωαννίνων

Περίληψη

Η παρούσα εργασία ασχολήθηκε με τη σύνθεση και το χαρακτηρισμό σύνθετων μεταλλικών υλικών μήτρας αλουμινίου με ενίσχυση δύο ενδομεταλλικών ενώσεων ΑΙ9Co2 και ΑΙ13Co4. Με την μέθοδο της κονεομεταλλουργίας παρασκευάστηκε το σύνθετο υλικό σε τρεις συστάσεις 2, 5 και 10 %vol. Τα δοκίμια υπέστησαν φθορά ολίσθησης, διάβρωση σε θαλασσινό περιβάλλον και καταγραφή των τιμών της μακροσκληρότητας. Τέλος, εξετάστηκαν στο ηλεκτρονικό μικροσκόπιο σάρωσης τα ίχνη φθοράς και τα ψήγματα (debris) καθώς και οι τομές διάβρωσης. Τα δοκίμια παρουσίασαν ομοιόμορφη διασπορά κατά μήκος του υλικού. Από τις πανοραμικές φωτογραφίες του μικροσκοπίου παρατηρήθηκε η ύπαρξη συσσωματωμάτων, μερικών πόρων και οξειδίων, πράγμα αναμενόμενο για τη συγκεκριμένη κονεομεταλλουργική οδό που χρησιμοποιήθηκε. Τα δοκίμια με ενίσχυση 2% vol και 10% vol παρουσίασαν ικανοποιητική πυροσυσσωμάτωση των σωματιδίων και ομοιόμορφη, συνεκτική μικροδομή γεγονός που ευνοεί τη συμπεριφορά σε φθορά και σε υδατική διάβρωση ενώ εξασφάλισε ικανοποιητικές τιμές μακροσκληρότητας. Αξίζει να σημειωθεί ότι με την αύξηση του ποσοστού ενίσχυσης στη μήτρα του σύνθετου υλικού, παρατηρούνται περισσότερα συσσωματώματα και πόροι. Παρόλο αυτά, λόγω της ομοιομορφίας της δομής τα φαινόμενο αυτά δεν επηρεάζουν σε μεγάλο βαθμό την επίδοση των δύο δοκιμίων (2 και 10%vol). Στο σύνθετο υλικό Al- 5%vol (Al9Co2, All3Co4), παρατηρήθηκαν σημεία τα οποία δεν έχουν πυροσυσσωματωθεί επαρκώς, το οποίο αποτέλεσε αρνητικό παράγοντα για την ανάπτυξη των επιθυμητών ιδιοτήτων του υλικού. Τέλος, στα πλαίσια του πειραματικού ενδιαφέροντος, σε μερικά δείγματα προστέθηκε μείγμα τολουολίου με πολύ-βουταδιένιο κατά την διαδικασία τη ανάμειξης με στόχο την καλύτερη πάκτωση των κόνεων κατά τη συμπίεση. Η συγκεκριμένη δοκιμή δεν είχε επιτυχία μιας και δημιουργήθηκαν μεγαλύτερα.

συσσωματώματα κατά τη συμπίεση.